

***N*-(2,5-Dichlorophenyl)-*N'*-[(*E*)-3-thienylmethylene]hydrazine**Vijayakumar N. Sonar,<sup>a</sup> Sean Parkin<sup>b</sup> and Peter A. Crooks<sup>a\*</sup><sup>a</sup>Department of Pharmaceutical Sciences, College of Pharmacy, University of Kentucky, Lexington, KY 40536, USA, and <sup>b</sup>Department of Chemistry, University of Kentucky, Lexington, KY 40506, USA

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**Key indicators**Single-crystal X-ray study  
*T* = 90 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$   
*R* factor = 0.038  
*wR* factor = 0.109  
Data-to-parameter ratio = 18.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{11}\text{H}_8\text{Cl}_2\text{N}_2\text{S}$ , was prepared from the condensation reaction of thiophene-3-carbaldehyde with 2,5-dichlorophenylhydrazine. The azomethine  $\text{C}=\text{N}$  bond has *E* geometry. No thienyl ring flip was observed.

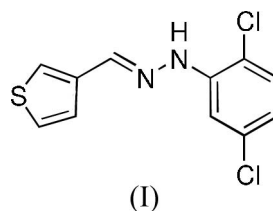
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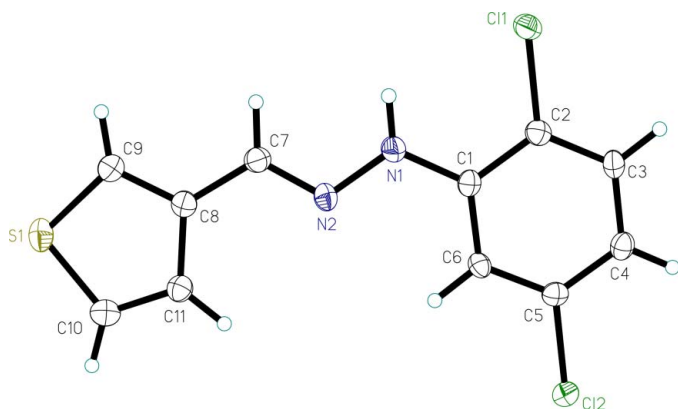
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**Comment**

After a long period during which tuberculosis seemed to be declining, the last two decades have seen an unexpected return (Raviglione *et al.*, 1995). Numerous recent reports in the literature provide evidence of renewed interest as the search for new antitubercular agents continues. A series of novel hydrazones have been synthesized and screened for antitubercular activity at the Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF), and the title compound, (I), showed 70% growth inhibition in the preliminary screen against *Mycobacterium tuberculosis* at  $6.25 \mu\text{g ml}^{-1}$ . The title compound was prepared from the condensation reaction of thiophene-3-carboxaldehyde with 2,5-dichlorophenylhydrazine, which afforded a single geometrical isomer. In order to confirm the  $\text{C}=\text{N}$  double-bond geometry of this compound, and to obtain more detailed information on the structural conformation of the molecule, its X-ray structure determination has been carried out and the results are presented here.



The molecular structure and atom-numbering scheme of (I) are shown in Fig. 1, and selected bond lengths and angles are listed in Table 1. In the molecule of (I), the  $\text{C}=\text{N}$  bond connecting the 2,5-dichlorophenylamino and thiophene rings has *E* geometry, with the two rings on opposite sides of the  $\text{C7}=\text{N2}$  bond. The molecule is nearly planar, with an  $\text{N1}-\text{N2}=\text{C7}-\text{C8}$  torsion angle of  $178.71(18)^\circ$  (Fig. 1). The  $\text{N1}-\text{N2}$  bond distance of  $1.368(2) \text{ \AA}$  is shorter than a normal  $\text{N}-\text{N}$  single bond, as in the case of 2,4-dinitrophenylhydrazine [ $1.405(6) \text{ \AA}$ ; Okabe *et al.*, 1993], suggesting delocalization of the azomethine double bond with the thiophene ring. This observation is further supported by the shortening of the  $\text{C7}-\text{C8}$  bond [ $1.445(3) \text{ \AA}$ ] compared with the standard value for a  $\text{C}_{\text{ar}}-\text{C}_{\text{sp}^2}$  single bond (Allen *et al.*, 1992). Also, the  $\text{N1}-\text{C1}$  bond distance of  $1.383(3) \text{ \AA}$  (Table 1) indicates partial


**Figure 1**

A view of the molecule (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.

double-bond character between atom N1 and atom C1 of the 2,5-dichlorophenyl ring.

The mode of packing of (I) along the *a* direction is illustrated in Fig. 2. Intermolecular hydrogen bonding exists between the imino H and the Cl atoms (Table 2), and this contributes to the stabilization of the crystal structure.

## Experimental

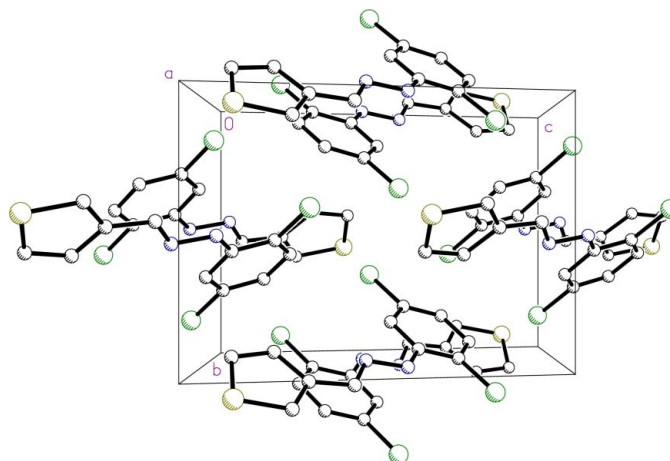
A mixture of thiophene-3-carboxaldehyde (0.336 g, 3 mmol) and 2,5-dichlorophenylhydrazine (0.531 g, 3 mmol) was dissolved in methanol (10 ml) and the solution was refluxed for 2 h. After cooling the reaction mixture, crystals of (I) formed and were collected by filtration. Crystallization from methanol gave (I) as a brown blocks which were suitable for X-ray analysis.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 6.75 (*dd*, 1H), 7.17 (*d*, 1H), 7.35 (*q*, 1H), 7.43 (*dd*, 1H), 7.55 (*m*, 2H), 7.89 (*s*, 2H);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ,  $\delta$ , p.p.m.): 114.1, 115.1, 119.8, 124.9, 125.2, 126.7, 130.0, 134.0, 136.3, 137.7, 141.4.

### Crystal data

$\text{C}_{11}\text{H}_8\text{Cl}_2\text{N}_2\text{S}$	$D_x = 1.562 \text{ Mg m}^{-3}$
$M_r = 271.15$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 2826 reflections
$a = 12.3563 (2) \text{ \AA}$	$\theta = 1.0\text{--}27.5^\circ$
$b = 8.3200 (2) \text{ \AA}$	$\mu = 0.71 \text{ mm}^{-1}$
$c = 11.6818 (3) \text{ \AA}$	$T = 90.0 (2) \text{ K}$
$\beta = 106.2747 (10)^\circ$	Block, brown
$V = 1152.82 (5) \text{ \AA}^3$	$0.20 \times 0.20 \times 0.15 \text{ mm}$
$Z = 4$	

### Data collection

Nonius KappaCCD area-detector diffractometer	2640 independent reflections
$\omega$ scans	2054 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SCALEPACK; Otwinowski & Minor, 1997)	$R_{\text{int}} = 0.027$
$T_{\text{min}} = 0.870$ , $T_{\text{max}} = 0.900$	$\theta_{\text{max}} = 27.5^\circ$
5066 measured reflections	$h = -16 \rightarrow 15$
	$k = -10 \rightarrow 10$
	$l = -15 \rightarrow 15$


**Figure 2**

A packing diagram for (I), viewed down the *a* axis. H atoms have been omitted.

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.038$   
 $wR(F^2) = 0.109$   
 $S = 1.09$   
 2640 reflections  
 145 parameters  
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0632P)^2 + 0.1691P]$$

where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.002$   
 $\Delta\rho_{\text{max}} = 0.57 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.47 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

N1—N2	1.368 (2)	S1—C9	1.705 (2)
N1—C1	1.383 (3)	S1—C10	1.709 (2)
N2—C7	1.283 (3)	C7—C8	1.445 (3)
N2—N1—C1	119.83 (17)	N1—C1—C2	119.38 (18)
C7—N2—N1	116.31 (18)	N2—C7—C8	120.8 (2)
C9—S1—C10	92.38 (11)	C9—C8—C7	123.8 (2)
C1—N1—N2—C7	−168.53 (19)	N1—N2—C7—C8	178.71 (18)
N2—N1—C1—C6	−5.6 (3)	N2—C7—C8—C9	−178.1 (2)
N2—N1—C1—C2	175.71 (18)	N2—C7—C8—C11	−1.8 (3)

**Table 2**

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
N1—H1 $\cdots$ Cl1	0.88	2.52	2.9395 (17)	110

H atoms were positioned geometrically and treated as riding, with  $N\text{—}H = 0.88 \text{ \AA}$  and  $C\text{—}H = 0.95 \text{ \AA}$ , and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ .

Data collection: COLLECT (Nonius, 1999); cell refinement: SCALEPACK (Otwinowski & Minor, 1997); data reduction: DENZO-SMN (Otwinowski & Minor, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL (Sheldrick, 1995); software used to prepare material for publication: SHELXL97 and local procedures.

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out by the Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF) at the National Institute of Allergy and Infectious Disease, Southern Research Institute, GWL Hansens Disease Center and Colorado State University, USA.

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